

# CORRELATING SUPRAMOLECULAR BEHAVIOR IN BUFFER AND PLASMA

VERSION 1.1    Nadia Mirza, Senior R&D and application Scientist, Fidabio

**Key Benefits of Fidabio determining the transitions between various structures in buffer and plasma**

- **Rapid and efficient molecular characterisation directly in buffer and plasma**
- **Quality data for very low concentrations of analyte - not possible on any other technology**
- **Data corresponds with SAXS**
- **Low amount of sample volume**

# INTRODUCTION

The present application note is based on work done in collaboration with Nikokaj Riis Christensen, PhD, et al. at the University of Copenhagen. A potential therapeutic molecule (arbitrarily named mX) was characterised for its properties in aqueous buffer as well as in human plasma. Due to its chemical structure, mX can self-assemble into higher order oligomeric structures, thus transitioning between monomer, oligomer and micelle structures. It was shown that Fida 1 provided reliable data for determining the transitions between various structures and good quality data for very low concentrations of mX, which was not possible on any other technology. In addition, FIDA experiments elucidated plasma behaviour of mX and showed it's binding to HSA.

# MATERIAL & METHODS

Fida 1 instrument with 480 nm LED fluorescence detection was used for assay development (Fida Biosystems ApS) with Fida standard capillary (i.d.: 75  $\mu\text{m}$ , LT: 100 cm,  $L_{\text{eff}}$ : 84 cm). PBS buffer at pH 7.5 was used. mX was fluorescently labelled. All samples were prepared in buffer and loaded onto the Fida 1 480 system, using either single glass vials or using a 96-well plate setup. The Fida 1 capillary was coated with HScoting, BSA coating (19 mg/ml), or using Poly-Ornithine (1 mg/ml), otherwise an uncoated capillary was used. Samples were run in one of two ways; after flushing of the capillary with buffer, the capillary was filled with analyte (unlabelled) or buffer using 1500 mbar for 45 s, followed by a single injection of sample (indicator) using 50 mbar for 10 s. The sample was then flushed through using 400 mbar for 180 s.

# RESULTS

## In solution behavior of mX and comparison with SAXS

In solution, mX displays an interesting behaviour. Given its chemical structure, mX self-assembles into higher order oligomeric structures with a hydrophobic part being encapsulated by a hydrophilic part (micelle). Earlier investigations using SAXS found that mX assembles into micelles with a radius of gyration ( $R_g$ ) of  $\sim 23\text{\AA}$  (Figure 1A, 1B)

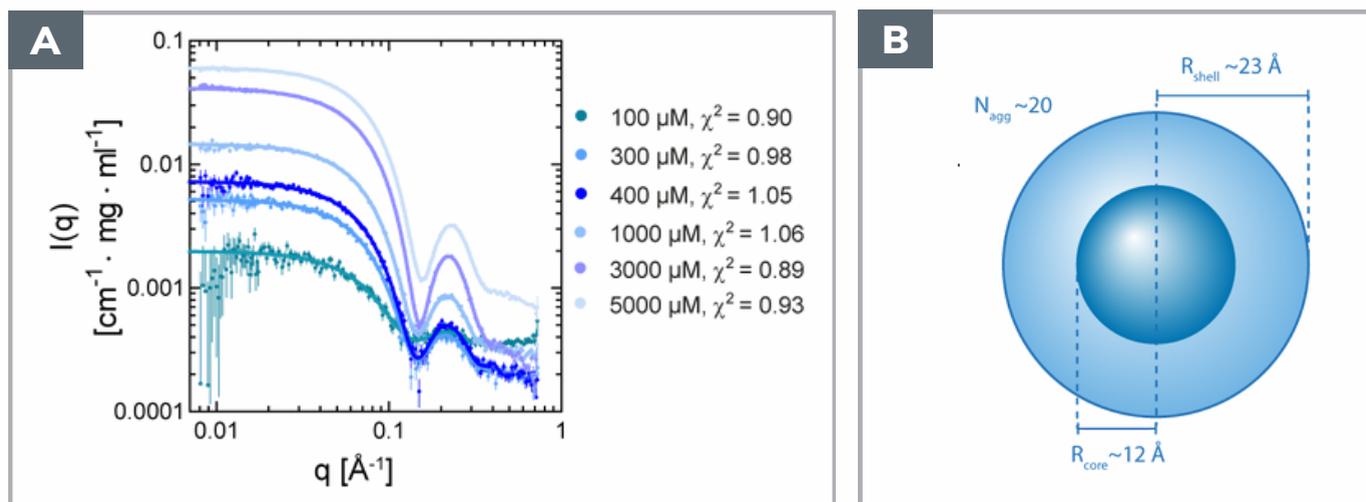


Figure 1: A) Small angle X-ray scattering of mX confirms self-assembly of mX into micelle structures. B) Best fit core shell model features, where  $N_{agg}$  is the number of subunits making up the average micelle,  $R_{core}$  is the radius of the core region containing the lipid tails, and  $R_{shell}$  is the radius of the shell, corresponding to  $R_g$ .

SAXS could, however, not provide information on critical micelle concentration (CMC) of mX. For better estimation of the CMC and the size of the assembled oligomer of mX, Fida 1 was used. In the experiments, a tracer of mX was used in combination with the Fida 1 system equipped with a fluorescent detector. This approach gave good quality data down to low  $\mu\text{M}$  concentrations and a CMC value was obtained by using a combination of FIDA capillaries and running a range of mX concentrations (Figure 2).

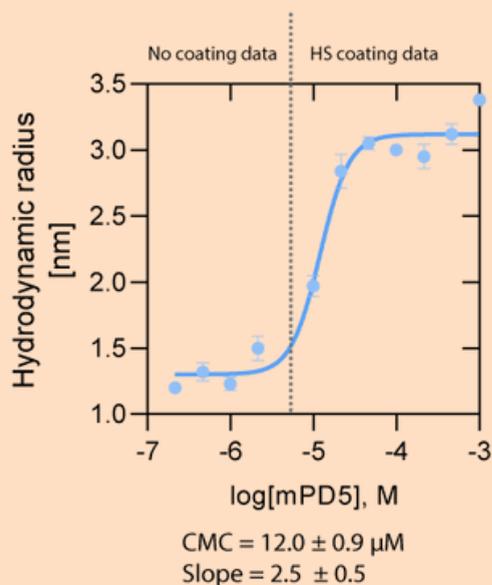


Figure 2: Binding isotherm of mX combining optimal coatings for different concentrations results in a CMC of  $12 \mu\text{M}$ . Data was plotted using GraphPad Prism 8.3, and raw data was fitted using Fida Software (2.0) and resulting hydrodynamic radii was plotted and fitted using a four parameters saturation binding curve with variable slope.

### Plasma behavior of mX

As a potential therapeutic compound, mX requires subcutaneous or intravenous administration, and therefore the properties of mX in plasma are essential. Two outcomes were envisioned; mX would remain in a micelle assembly and thereby obtain plasma stability, or mX could alternatively be stabilised by binding to plasma proteins, such as Human serum albumin (HSA). We therefore tested; 1) the ability of mX to bind to HSA in solution and 2) the properties of mX in human plasma to validate either presence of micelles or HSA complexes in a more complex sample matrix. It was found that mX could indeed bind to HSA (Figure 3). Furthermore, the dominant species in plasma had an Rh of  $\sim 42\text{-}47$  Å suggesting that mX associates with HSA rather than forming micelles in plasma.

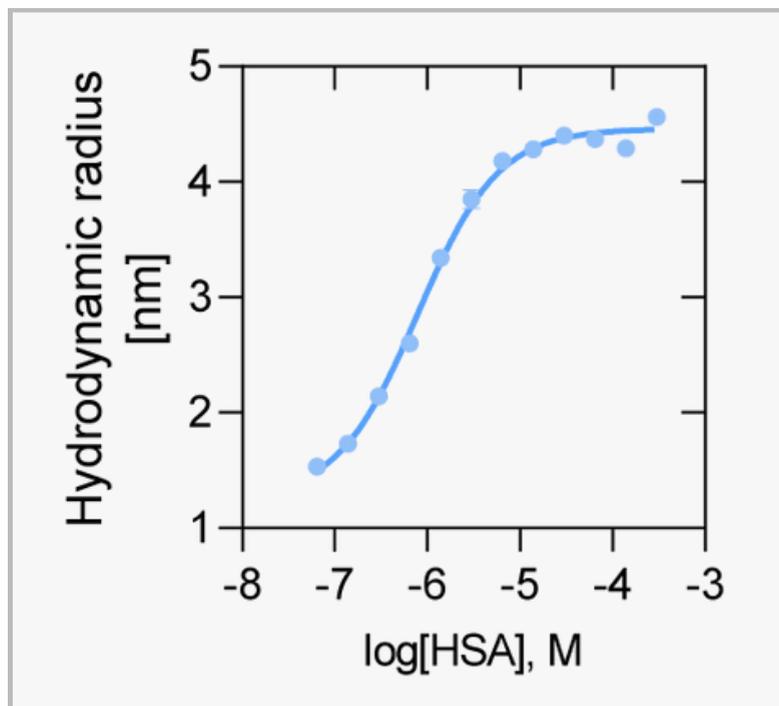


Figure 3: Binding isotherm of mX binding to HSA suggests an affinity (KD) of 787 nM and a hydrodynamic radius (RH) of the complex of 4.46 nm.

## CONCLUSION

Fida 1 enables quick and easy validation of mX micelle formation with an Rh of  $\sim 30$  Å, which is in good agreement with data obtained from SAXS. In addition, FIDA setup was able to provide a CMC value of 12  $\mu\text{M}$  for mX, a critical information not obtained by any other techniques. It was also shown that in human plasma, mX could bind to HSA with a sub  $\mu\text{M}$  affinity. FIDA, thereby effectively elucidated mX behaviour in solution and in plasma, an important information for a potential therapeutic compound.